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Particle and Crystallite Size Characterisation of Lead Titanate Derived from Solid-state Reaction Method

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ABSTRACT: Lead titanate (PbTiO_3) ceramic was produced via solid state reaction via a vibratory ball milling machine and subsequent heat treatment. The effect of milling time on the particle and crystallite size of PbTiO_3 powder was investigated. Powder samples were studied using particle size analyser (PSA). The annealing process was up to $1,000^\circ\text{C}$ and the products were examined by X-ray diffractometer (XRD) to determine phase formation and crystallite size. It was found that the average particle size of powder initially increased due to laminated layers formation and then decreased to an asymptotic value of $\sim 0.8 \mu\text{m}$ as the milling time extended even to a relatively longer time. Single-phase PbTiO_3 were achieved at 600°C for 1 h holding time of annealing temperature. Annealing the sample of the particles at $1,000^\circ\text{C}$ resulted in a dense compact and promoted the formation of particles containing nanocrystallites. The crystallite size of PbTiO_3 increased as the function of temperature of annealing process.

Keywords: Lead titanate, particle size, crystallite size, solid state reaction, annealing

1. INTRODUCTION

Ferroelectrics are materials with reversible spontaneous polarisation.¹⁻⁴ Lead titanate (PbTiO_3) is one of the fundamental ferroelectric materials with an ABO_3 compound structure called perovskite and the highest spontaneous polarisation among all the ferroelectric perovskites.⁵⁻¹⁴ According to the first-principle calculations on ferroelectric perovskites, hybridisation between the electronic

states of A and B atoms and the oxygen atoms is essential for ferroelectricity. PbTiO_3 has highest tetragonal distortion ($c/a \approx 1.063$) among all members of the perovskite's family. This tetragonal distortion corresponds to the highest spontaneous polarisation among all the ferroelectric perovskites. Perovskite-type PbTiO_3 has a high spontaneous polarisation of $86 \mu\text{C cm}^{-2}$, Curie temperature of 364°C – 490°C , a relatively low permittivity, a large piezoelectric coefficient ($250 \mu\text{C cm}^{-2} \text{K}^{-1}$) and small dielectric constant.^{5,10,15–17} The dielectric constant increased in the annealing range of 450°C – 750°C . This trend is due to the increased grain size and higher crystallinity with annealing temperatures.¹⁸ The values of saturation polarisation (P_s), remanent polarisation (P_r) and coercive field (E_c) PbTiO_3 are $13.1 \mu\text{C cm}^{-2}$, $3.2 \mu\text{C cm}^{-2}$ and 5.1 kV cm^{-1} , respectively. Because of their character, ferroelectrics are widely used in many applications: the ferroelectric random-access memory field, pyroelectric infrared sensor, electro-optic devices, insulator gates in metal-insulator-semiconductor diodes, capacitors, transistor, piezoelectric actuators, high frequency ultrasonic transducers and so on.^{19–22}

The PbTiO_3 property within its applications depends on several aspects: the purity of materials indicated in stoichiometry number; and microstructure that consists of phase, particle's size and then crystallite's size. To obtain the stoichiometry, particle and crystallite size PbTiO_3 , different preparation methods have been introduced such as co-precipitation, emulsion or hydrothermal treatment sol-gel and spark plasma sintering sonochemical, besides the conventional solid-state reaction of mixed oxides or mechanical alloying.^{4,6,9–12,16,20,23–35} All methods will produce varying microstructures, processes and then manufacture temperature. The processing time must be determined for quick process and the better quality of final product; it has higher purity in single phase, and the particle and crystallite formed in nanometer size. Chattopadhyay et al. conducted a detailed study on the influence of particle size on the ferroelectric properties of lead titanate.³⁶ Studies revealed that size effects are important only below 100 nm. The usual method for producing the fine nanocrystalline materials is mechanical alloying and milling by a ball-milling technique, which has also been adapted into the preparation of lead titanite. The technique is considered simple and less costly for producing very fine particles. In this study, mechanical alloying and milling method has been developed to produce the fine nanocrystalline materials of PbTiO_3 ceramics. Finally, the results were assessed on material characterisation consisting of particle and crystallite size.

2. EXPERIMENTAL

PbTiO₃ was obtained from the mixture of lead (II) carbonate (PbCO₃) and titanium (IV) oxide (TiO₂) powders by using high-energy ball milling and heat treatment processes. Stoichiometric quantities of the analytical-graded precursors PbCO₃ and TiO₂ with purities better than 98% were mixed and milled in a vibratory ball mill up to 10 h. The weight ratio of balls to milled material was 1:10. After milling process, the diameter sizes of examined powder particles were determined using particle size analyser (PSA) Mastersizer ZS Nanoseries. Phase analysis and crystallite size of milled powders were carried out using the X-ray Philips diffractometer equipped with CuK α radiation. The X-ray diffraction (XRD) patterns were recorded by “step-scanning” method. The powders were annealed in the electric chamber furnace (Nabertherm N31/H) at 500°C, 600°C, 800°C, 900°C and 1,000°C in the air under atmosphere pressure up to 1 h. The Rietveld analysis was performed applying High Score Plus program that is an updated version for Rietveld refinement with PC and mainframe computers. The pseudo-Voigt function was used in describing the diffraction line profiles at Rietveld refinement. The crystallite sizes for PbCO₃, TiO₂ and also PbTiO₃ phases were estimated using Williamson-Hall method.³⁷ Intensity data during scanning of 2 θ was taken for each step of the diffraction angle 0.005°. Diffraction peak width (B) is given by Equation 1 and the mean crystallite size (D) obtained from Equation 2:

$$B = \frac{0.9\lambda}{D \cos \theta} + \eta \tan \theta \quad (1)$$

$$B \cos \theta = \frac{0.9\lambda}{D} + \eta \sin \theta \quad (2)$$

where λ is the X-ray wavelength, η is the strain in the materials and θ is the Bragg angle, while the peak width B is obtained after the correction due to instrument broadening according to Equation 3:

$$B = \sqrt{B_o^2 - B_s^2} \quad (3)$$

B_o is the Full Width at Half Maximum (FWHM) of the test sample, and B_s is the FWHM standard samples that used silicon (Si).

3. ²³ RESULTS AND DISCUSSION

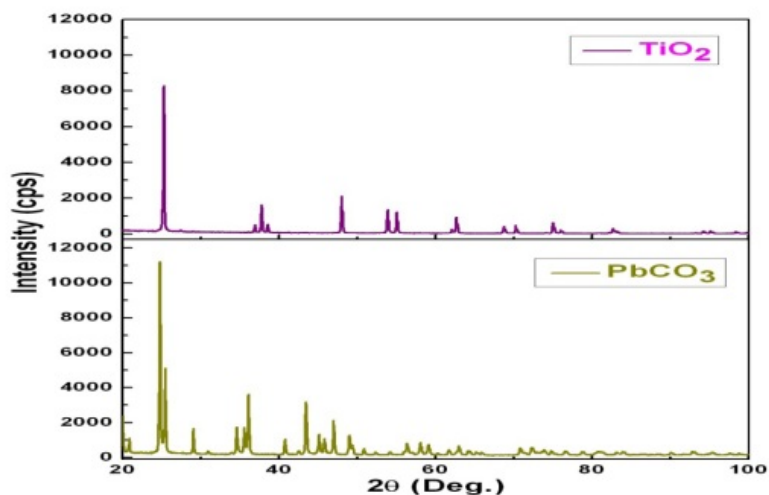


Figure 1: XRD patterns of TiO_2 and PbCO_3 precursors.

The diffraction patterns of TiO_2 and PbCO_3 precursors shown in Figure 1 which were matched with diffraction patterns of TiO_2 and PbCO_3 in Inorganic Crystal Structure Database (ICSD) number 98-009-6946 and 98-016-6089 respectively. Based on the Rietveld analysis, the crystal system of TiO_2 and PbCO_3 are tetragonal and orthorhombic respectively.

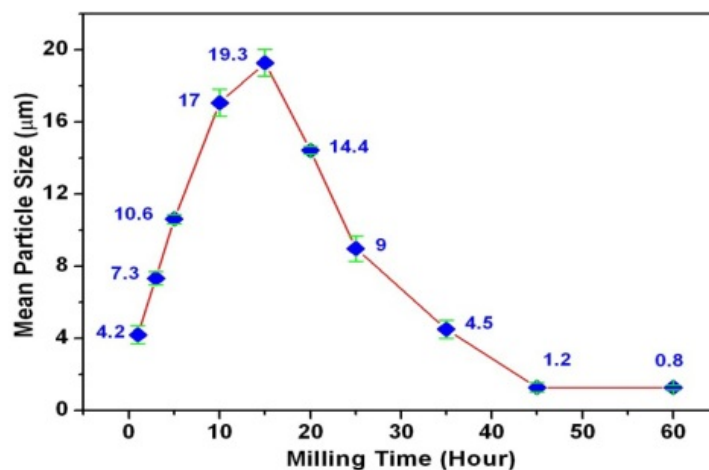


Figure 2: The mean particle size of TiO_2 and PbCO_3 mixture with milling time.

Figure 2 shows results of evaluation for mean particle size of TiO_2 and PbCO_3 mixture up to 60 h of milling. All sample powders went through the four stages of the mechanical alloying process, namely: (a) initial stage; (b) intermediate stage; (c) final stage; and (d) completion stage.³⁸ It shows that the mean particle sizes of mechanically milled TiO_2 and PbCO_3 mixture in initial or early stages of milling are characterised by the increase in the mean particle size due to incorporation of particles of component compounds. The mean particle size of the material increased from 4.2 μm to 17 μm at duration 1 h to 10 h of mixing. The core compounds experienced cold weld, namely the integration of the two particles of the basic compounds to form a close bond between the particles as a consequence of the ball mill impact. The process of impact between ball mills continuously occurred. The largest mean particle size were achieved after 15 h milling times, while the mean particle size of TiO_2 and PbCO_3 mixture is $\sim 19 \mu\text{m}$. As the milling time extended beyond 15 h, the mean size towards a settled value decreased progressively. Long terms of mechanical treatment during advanced stages of mechanical alloying caused the particles to experience embrittlement due to accumulation of internal stresses.³⁹ Continuous plastic deformations of the brittle particles caused further reduction in particle size towards an average value of $\sim 0.8 \mu\text{m}$ and eventually settled down to that value even when the deformation continued to grow after the duration of 60 h milling time at completion stage.

The comparison of diffraction patterns of TiO_2 and PbCO_3 mixture after 1 h, 5 h, 10 h, 25 h, 40 h and 60 h of milling process is shown in Figure 3. Identification of the diffraction peaks ensured that all peaks matched with that of TiO_2 and PbCO_3 phase.

Figure 4 illustrates the evaluation of mean crystallite size in milled particles after milling process based on XRD patterns using Williamson-Hall method. Figure 4 shows the ball milling process for 60 h in a mixture of TiO_2 and PbCO_3 lead to a decline in the value of the average crystallite size. The average size of the TiO_2 crystallites decreased not so significantly exponentially with time. In contrast to TiO_2 , after mechanical alloying of 60 h, the crystallite size of PbCO_3 decreased significantly about 1.5 times smaller. It shows that the process of milling up to 60 h resulted in a more brittle and easily shattered PbCO_3 when compared with TiO_2 . The mechanical milling process caused a decrease in the crystallite size of tested phases and led to homogenising of the milled mixture. The mean crystallite size of TiO_2 and PbCO_3 phases diminished to 87 nm and 80 nm, respectively of the milling time up to 60 h.

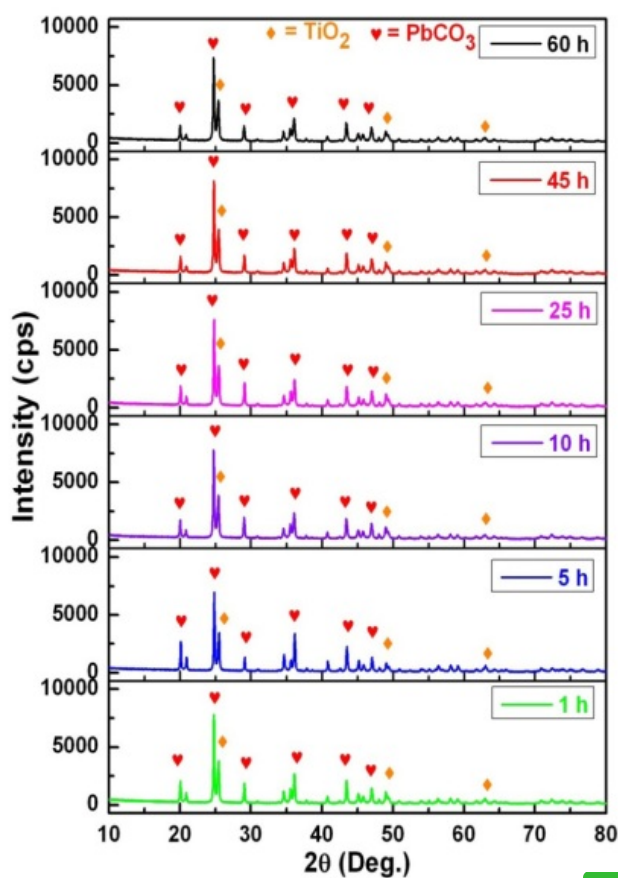


Figure 3: XRD patterns of TiO_2 and PbCO_3 mixture up to 60 h of milling.

The XRD investigations of TiO_2 and PbCO_3 powder mixture milled for 60 h and after different temperature and times of annealing treatment up to $1,000^\circ\text{C}$ are shown in Figure 5.

At 500°C with a holding time of 1 h, the single phase PbTiO_3 has not yet been formed where there is still another phase present, $\text{Pb}_5\text{C}_3\text{H}_2\text{O}_{12}$. Single phase PbTiO_3 with tetragonal perovskite crystal structure was formed after annealing at 600°C up to $1,000^\circ\text{C}$ in 1 h. XRD pattern of the sample which has undergone annealing process at 600°C for 1 h matched with the data based on the diffraction pattern on ICSD number 98-009-0693. The lattice constant calculated from the XRD data is $a = b = 3.9116 \text{ \AA}$ and $c = 4.0943 \text{ \AA}$. The structure matches with the PbTiO_3 that can be used in various applications. Based on the XRD patterns in Figure 5, the average size of crystallites of each phase can be found, where the result is shown in Figure 6.

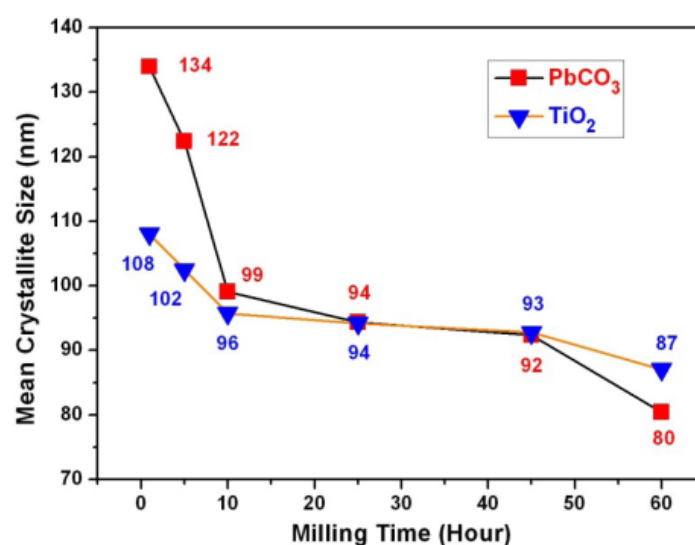


Figure 4: The mean crystallite size of TiO₂ and PbCO₃ mixture up to 60 h of milling.

To support the result, the temperature and time of the most optimum were achieved in the transformation process into a single phase PbTiO₃ phase in full, then tested with thermal analysis. The intensity and sharpness of the XRD peaks of PbTiO₃ phase were found to increase with annealing temperature at 600°C up to 1,000°C. Based on Figure 6, it can be concluded that the average size of crystallites PbTiO₃ increased exponentially with the rising annealing temperature. The increase in intensity and sharpness of the XRD peaks with annealing temperature may be attributed to the increase in the grain or crystallite size thereby increasing packing density of the samples annealed at higher temperature.

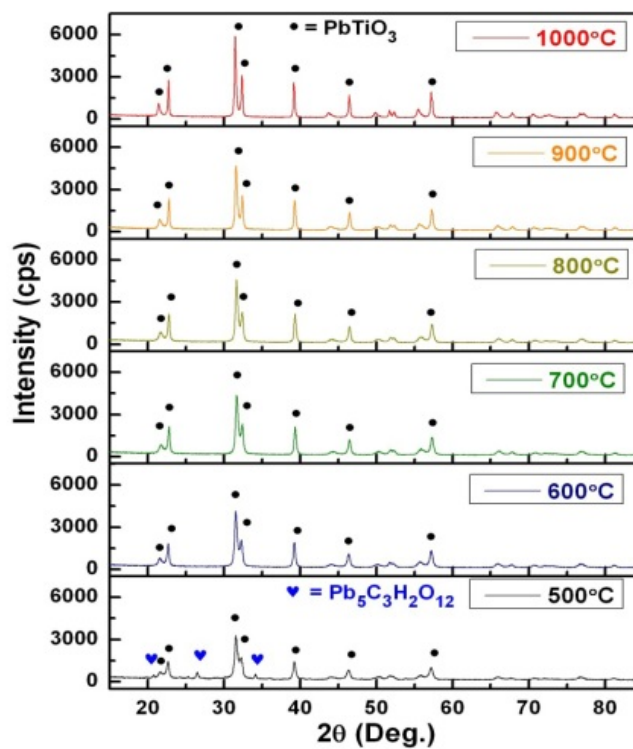


Figure 5: XRD profile of TiO_2 and PbCO_3 mixture after annealing up to 1,000°C.

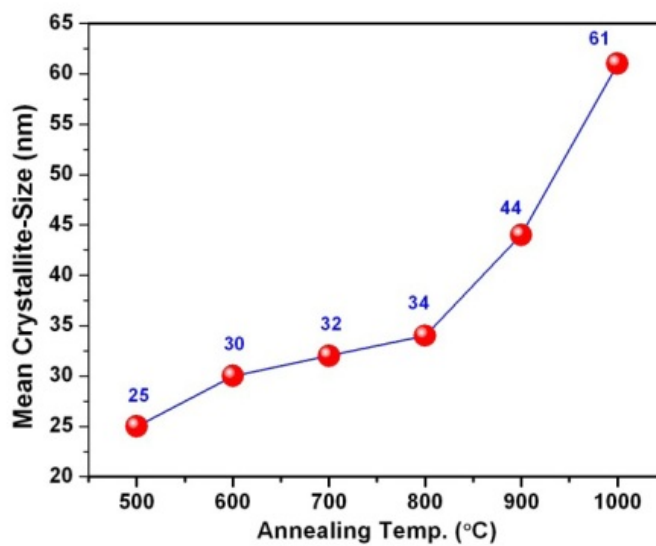


Figure 6: The mean crystallite size of PbTiO_3 .

4. CONCLUSION

Tests on a mixture of TiO_2 and PbCO_3 as a piezoelectric material, PbTiO_3 , after undergoing a process of milling and sintering, concluded that mechanical alloying process for 60 h in a mixture of $\text{TiO}_2 + \text{PbCO}_3$ caused the mixture of the two compounds to decrease their average particle size to $0.8 \mu\text{m}$ and a crystallite size of 87 nm and 80 nm, respectively. The reduction of the size of particle and crystallite was the result of the continuous collision between sample powder and ball mill. As a consequence, the samples underwent embrittlement and deformation. Another conclusion that can be reported is that PbTiO_3 ceramic has been prepared by conventional solid-state reaction processing technique. The as-fired powder was found to be amorphous and crystallised to tetragonal PbTiO_3 after annealing at 600°C up to $1,000^\circ\text{C}$ for 1 h. The crystallite size of PbTiO_3 increases as a function of temperature of annealing process.

5. ACKNOWLEDGEMENTS

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6. REFERENCES

1. Gene, H. H. (1999). Ferroelectric ceramics: History and technology. *J. Am. Ceram. Soc.*, 82(4), 797–818, <https://doi.org/10.1111/j.1151-2916.1999.tb01840.x>.
2. Cohen, R. E. (1992). Origin of ferroelectricity in perovskite oxides. *Nature*, 358(6382), 136–138.
3. Pardo, L. et al. (1999). Ferroelectric materials based on lead titanate. In Nalwa, H. S. (Ed). *Handbook of low and high dielectric constant materials and their applications*. Jakarta: Academic Press.
4. Ricote, J. et al. (2004). Application of the X-ray combined analysis to the study of lead titanate based ferroelectric thin films. *Thin Solid Films*, 450(1), 128–133, <https://doi.org/10.1016/j.tsf.2003.10.056>.
5. Bhatti, H. S. et al. (2016). Synthesis and induced multiferroicity of perovskite PbTiO_3 : A review. *Appl. Surf. Sci.*, 367, 291–306, <https://doi.org/10.1016/j.apsusc.2016.01.164>.

6. Wang, D. G. et al. (2008). Lead-based titanate ferroelectric thin films fabricated by a sol-gel technique. *Appl. Surf. Sci.*, 255, 1637–1645, <https://doi.org/10.1016/j.apsusc.2008.09.053>.
7. Vijendra, A. C. & Govind, K. B. (2013). Synthesis, structure and electrical properties of pure PbTiO_3 ferroelectric ceramics. *Smart Mater. Res.*, 1–9, <http://dx.doi.org/10.1155/2013/147524>.
8. Ahmad, T. & Ashok, K. G. (2006). Reverse micellar route to nanocrystalline titanate (SrTiO_3 , SrTiO_4 , and PbTiO_3): Structural aspects and dielectric properties. *J. Am. Ceram. Soc.*, 89(4), 1326–1332, <https://doi.org/10.1111/j.1551-2916.2005.00886.x>.
9. Fang, J. et al. (2002). Comparative study on phase development of lead titanate powders. *Mater. Lett.*, 52, 304–312, [https://doi.org/10.1016/S0167-577X\(01\)00411-6](https://doi.org/10.1016/S0167-577X(01)00411-6).
10. Fang, J. et al. (1999). Preparation and characterisation of ultrafine lead titanate (PbTiO_3) powders. *J. Mater. Sci.*, 34, 1943–1952, <https://doi.org/10.1023/A:1004596217129>.
11. Jangade, P., Arjunwadkar, P. R. & Nagarawadi, M. A. (2016). Structural characterization of lead titanate (PbTiO_3) sample using FULLPROF. *IOSR J. Appl. Phys.*, 8(6), 52–60, <https://doi.org/10.9790/4861-0806055760>.
12. Kim, S., Jun, M. C. & Hwang, S. C. (1999). Preparation of undoped lead titanate ceramics via sol-gel processing. *J. Am. Ceram. Soc.*, 82(2), 289–296, <https://doi.org/10.1111/j.1551-2916.1999.tb20060.x>.
13. Zhu, J. et al. (2014). Pressure-induced reversal between thermal contraction and expansion in ferroelectric PbTiO_3 . *Scient. Rep.*, 4, 3700, <https://doi.org/10.1038/srep03700>.
14. Kvasov, A. et al. (2016). Piezoelectric enhancement under negative pressure. *Nature Comm.*, 7, 12136, <https://doi.org/10.1038/ncomms12136>.
15. Yan, Y. et al. (2016). Giant piezoelectric voltage coefficient in grain-oriented modified PbTiO_3 material. *Nature Comm.*, 7, 13089, <https://doi.org/10.1038/ncomms13089>.
16. Sverre, M. S. et al. (2007). Decomposition and crystallization of a sol-gel-derived PbTiO_3 precursor. *J. Am. Ceram. Soc.*, 90(8), 2649–2652, <https://doi.org/10.1111/j.1551-2916.2007.01789.x>.
17. Palkar, V. R., Purandare, S. C. & Pinto, R. (1999). Ferroelectric thin films of PbTiO_3 on silicon. *J. Phys. D Appl. Phys.*, 32, 1–18.
18. Yadav, H. O. (2004). Optical and electrical properties of sol-gel derived thin films of PbTiO_3 . *Ceram. Int.*, 30(7), 1493–1498, <https://doi.org/10.1016/j.ceramint.2003.12.133>.
19. Rodel, J. et al. (2009). Perspective on development of lead-free piezoceramics. *J. Am. Ceram. Soc.*, 92(6), 1153–1177.

20. Rujiwatra, A., Jongphiphan, J. & Ananta, S. (2005). Stoichiometric synthesis of tetragonal phase pure lead titanate under mild chemical conditions employing NaOH and KOH. *Mater. Lett.*, 59, 1871–1875, <https://doi.org/10.1016/j.matlet.2005.02.002>.
21. Zhang, S. & Yu, F. (2011). Piezoelectric materials for high temperature sensor. *J. Am. Ceram. Soc.*, 94(10), 3153–3170, <https://doi.org/10.1111/j.1551-2916.2011.04792.x>.
22. Zhang, S. et al. (2013). Piezoelectric property of relaxor-PbTiO₃ crystals under uniaxial transverse stress. *Appl. Phys. Lett.*, 102, 172902, <https://doi.org/10.1063/1.4803183>.
23. Gelabert, M. C., Gersten, B. L. & Riman, R. E. (2000). Hydrothermal synthesis of lead titanate from complexed precursor solutions. *J. Cryst. Growth*, 211(1–4), 497–500, [https://doi.org/10.1016/S0022-0248\(99\)00781-2](https://doi.org/10.1016/S0022-0248(99)00781-2).
24. Morita, T. & Cho, Y. (2006). Piezoelectric property of an epitaxial lead titanate thin film deposited by the hydrothermal method. *Appl. Phys. Lett.*, 88, 112908, <https://doi.org/10.1063/1.2183364>.
25. Zeng, X. et al. (2002). Preparation of nanocrystalline PbTiO₃ by accelerated sol-gel process. *Mater. Chem. Phys.*, 77, 209–214, [https://doi.org/10.1016/S0254-0584\(01\)00558-2](https://doi.org/10.1016/S0254-0584(01)00558-2).
26. Wu, Y. J. et al. (2010). Densification and microstructures of PbTiO₃ ceramics prepared by spark plasma sintering. *Mater. Sci. Eng. A*, 527, 5157–5160, <https://doi.org/10.1016/j.msea.2010.04.096>.
27. Wang, J. et al. (2005). Sonochemical preparation of PbTiO₃ fine powders. *J. Am. Ceram. Soc.*, 88(1), 34–37, <https://doi.org/10.1111/j.1551-2916.2004.00015.x>.
28. Udomporn, A. & Ananta, S. (2004). Effect of calcination condition on phase formation and particle size of lead titanate powders synthesized by the solid-state reaction. *Mater. Lett.*, 58, 1154–1159, <https://doi.org/10.1016/j.matlet.2003.08.028>.
29. Wongmaneerung, R., Yimnirun, R. & Ananta, S. (2006). Effects of milling time and calcination condition on phase formation and particle size of lead titanate nanopowders prepared by vibro-milling. *Mater. Lett.*, 60, 2666–2671, <https://doi.org/10.1016/j.matlet.2006.01.062>.
30. Wongmaneerung, R., Yimnirun, R. & Ananta, S. (2006). Effect of vibro-milling time on phase formation and particle size of lead titanate nanopowders. *Mater. Lett.*, 60, 1447–1452, <https://doi.org/10.1016/j.matlet.2005.11.043>.
31. Stankus, V. & Dudonis, J. (2004). Lead titanate thin film synthesis by solid-state reactions. *Mater. Sci. Eng. B*, 109, 178–182, <https://doi.org/10.1016/j.mseb.2003.10.047>.

32. Xue, J., Wan, D. & Wang, J. (1999). Mechanochemical synthesis of nanosized lead titanate powders from mixed oxides. *Mater. Lett.*, 39, 364–369, [https://doi.org/10.1016/S0167-577X\(99\)00036-1](https://doi.org/10.1016/S0167-577X(99)00036-1).
33. Somiya, Y., Bhalla, A. S. & Cross, L. E. (2001). Study of (Sr, Pb)TiO₃ ceramics on dielectric and physical properties. *Int. J. Inorg. Mater.*, 3, 709–714, [https://doi.org/10.1016/S1466-6049\(01\)00187-8](https://doi.org/10.1016/S1466-6049(01)00187-8).
34. Forrester, J. S. et al. (2004). Synthesis of PbTiO₃ ceramics using mechanical alloying and solid state sintering. *J. Solid State Chem.*, 177, 3553–3559, <https://doi.org/10.1016/j.jssc.2004.06.005>.
35. Udomporn, A. & Ananta, S. (2004). The phase formation of lead titanate powders prepared by solid-state reaction. *Curr. Appl. Phys.*, 4, 186–188, <https://doi.org/10.1016/j.cap.2003.11.005>.
36. Chattopadhyay, S. et al. (1995). Size-induced diffuse phase transition in the nanocrystalline ferroelectric PbTiO₃. *Phys. Rev. B*, 52, 13177, <https://doi.org/10.1103/PhysRevB.52.13177>.
37. Williamson, G. K. & Hall, W. H. (1953). X-ray line broadening from filed aluminium and wolfram. L'élargissement des raies de rayons x obtenues des limailles d'aluminium et de tungstène. Die verbreiterung der roentgeninterferenzlinien von aluminium- und wolframspänen. *Acta Metal.*, 1, 22–31, [https://doi.org/10.1016/0001-6160\(53\)90006-6](https://doi.org/10.1016/0001-6160(53)90006-6).
38. Lü, L. & Lai, M. O. (1998). *Mechanical alloying*, vol. 4. Boston: Kluwer.
39. Suryanarayana, C. (2004). *Mechanical alloying and milling*. New York: Marcel Dekker.

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- 29 N. Forouzanmehr, F. Karimzadeh, M.H. Enayati. "Synthesis and characterization of TiAl/ α -Al₂O₃ nanocomposite by mechanical alloying", Journal of Alloys and Compounds, 2009
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- 30 Tanguy Pussacq, Houria Kabbour, Silviu Colis, Hervé Vezin et al. "Reduction of Ln Ti O Layered Perovskites: A Survey of the Anionic Lattice, Electronic Features, and Potentials ", Chemistry of Materials, 2017
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- 33 "Structure investigations of the ground FeAl(Cr)-Cr₃C₂-Cr₇C₃-TiC composite powder prepared through the SHS process", International Journal of Microstructure and Materials Properties, 2007
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- 34 M. Tavoosi, F. Karimzadeh, M.H. Enayati.

"Fabrication of Al–Zn/ α -Al₂O₃ nanocomposite by mechanical alloying", Materials Letters, 2008

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Wisnu Ari Adi, A. Manaf. "Microstructure and Phase Analysis of La_{0.8}Ba_{0.2}Ti_xMn_(1-x)O₃ System for Microwave Absorber Material (x = 0 – 0.7)", Advanced Materials Research, 2013

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